

Note

HIGH TEMPERATURE DTA OF CERAMIC SUPERCONDUCTOR

$\text{YBa}_2\text{Cu}_3\text{O}_{7-\gamma}$

M NEVŘIVA, E POLLERT, J ŠESTÁK and A TRÍSKA

*Institute of Physics of the Czechoslovak Academy of Sciences, Na Slovance 2,
180 40 Praha 8 (Czechoslovakia)*

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During DTA of the $\text{YBa}_2\text{Cu}_3\text{O}_{7-\gamma}$ (1 2 3) ceramic oxide superconductor fired at 950°C for 50 h in a stream of oxygen four endothermic effects are detected within the range $950\text{--}1350^\circ\text{C}$. $\text{YBa}_2\text{Cu}_3\text{O}_{7-\gamma}$ decomposes reversibly to Y_2BaCuO_5 , BaCuO_2 and CuO at 999°C having a ternary eutectic at 957°C .

Ozawa and co-workers [1,2] found the formation of a perovskite phase in the temperature region above 750°C and ca 970°C they observed melting followed by decomposition and phase separation above this temperature. De Leeuw et al [3] reported a reversible decomposition of $\text{YBa}_2\text{Cu}_3\text{O}_{7-\gamma}$ above 1000°C in air into Y_2BaCuO_5 and CuO , and above 1200°C into Y_2O_3 , BaCu_2O_2 and CuO .

EXPERIMENTAL

For our high temperature thermal investigation we used a commercial DTA apparatus by Netzsch equipped by a DDK measuring head with Pt crucibles. The temperature was calibrated by ICTA/NBS temperature standards and Al_2O_3 was used as reference. A powdered sample of about 100 mg was heated in an oxygen and/or air stream at a rate of 5 K min^{-1} and temperature changes were detected with a sensitivity of $0.1\text{ mV}/25\text{ cm}$ by the thermocouple Pt–Pt10Rh. The 1 2 3 sample was prepared by mixing, powdering and firing of analysed oxides [4] finally tempered at 950°C for 50 h, followed by slow cooling to ambient temperature.

RESULTS AND DISCUSSION

Figure 1 shows the DTA curves for the 1 2 3 sample heated in oxygen up to 1350°C for the first and the second time. Four endothermic effects can be seen with a mutual intensity ratio of $14 : 100 : 3.5 : 3.4$ the last being

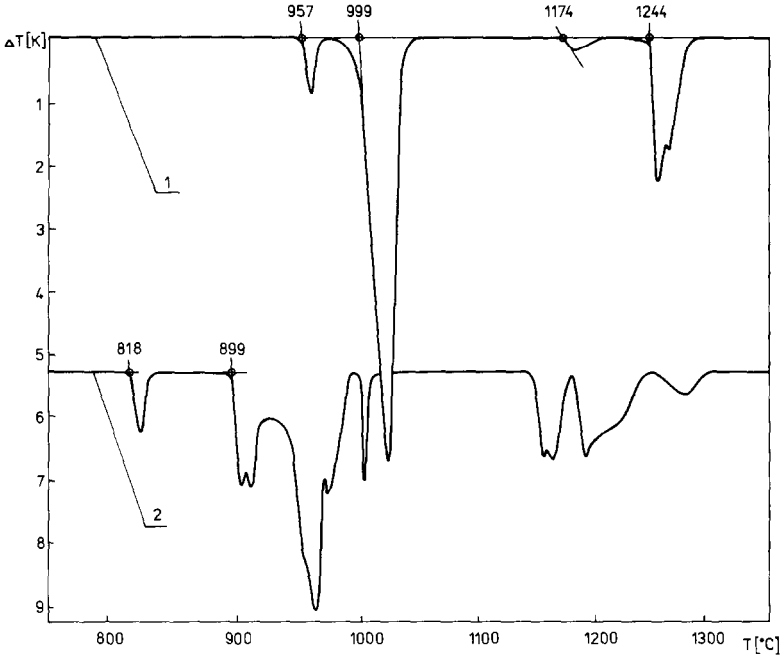


Fig 1 DTA curves of $\text{YBa}_2\text{Cu}_3\text{O}_{7-\gamma}$ in dynamic oxygen atmosphere (1) first run, (2) second run of the same sample

twisted By comparing with the DTA of known single phases of BaCuO_2 , YCuO_2 and Y_2BaCuO_5 we can deduce that the first effect is connected with an eutectic of the ternary $\text{YO}_{1.5}$ -BaO-CuO system

The second effect is associated with the decomposition of the 1 2 3 phase into Y_2BaCuO_5 and BaCuO_2 instantly dissociating into BaO and CuO, respectively The third is probably decomposition of the binary $\text{Y}_z\text{Ba}_y\text{O}_x$ phase which occurs together with the Y_2BaCuO_5 phase The last decomposes above 1244°C as is indicated by the last twisted peak The first decomposition was proved by an independent X-ray diffraction analysis on quenched samples If DTA is limited to below 1050°C the peak of the 1 2 3 phase decomposition is reversible and the sample remains superconductive

In contrast with the result of De Leeuw et al [3] we observed irreversible decomposition at temperatures above 1244°C During the second heating of the same sample the DTA pattern substantially changes due to the formation of new high temperature phases initiated by the formation of a liquid phase The eutectics at 818 and 900°C (twisted) are remarkable, these appear during cooling and remain during repeated runs In consequence the portion of the desired superconducting phase 1 2 3 decreases (more than four times with respect to the intensity ratios) because of the Y_2BaCuO_5 phase and the unknown phase formation It follows that the maximum

temperature for the 1 2 3 superconductor phase formation should not exceed the temperature of the second peak, i.e. 999°C which is in accordance with the temperature of 960°C recommended by Ozawa et al [1]. When carrying out the DTA in air the temperatures of the two first peaks decrease to 911 and 977°C respectively

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